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# Latest Study of Fe-Cr Based Filler Metals

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# Background

- In automobile industry, Ni based filler metals have been used for many applications to meet system requirements.
- Recent past history of soaring Ni price, high volume users of Ni based filler metals had seen dramatic increase in their brazing cost.
- In response to these changing market conditions, a new high strength, low cost alternative to traditional Ni based filler metals needed to be developed.

⇒ Fe-Cr based filler metal was developed.

# Background

## Service Conditions:

- High Operation Temperature
- High Thermal Stress
- Oxidative Environment
- Corrosive Environment

## Structures:

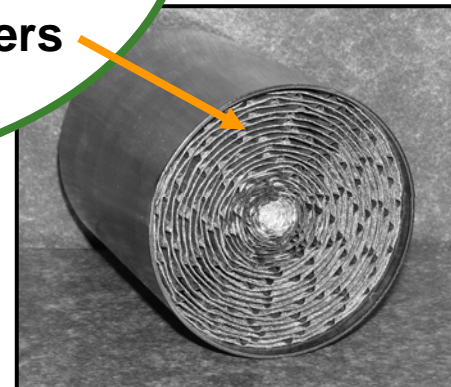
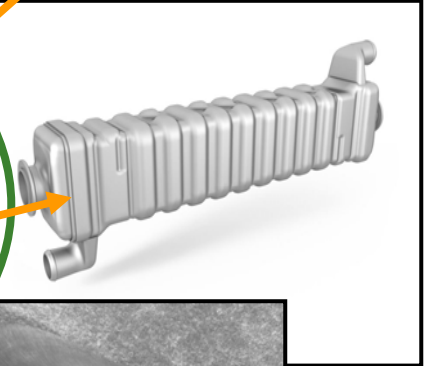
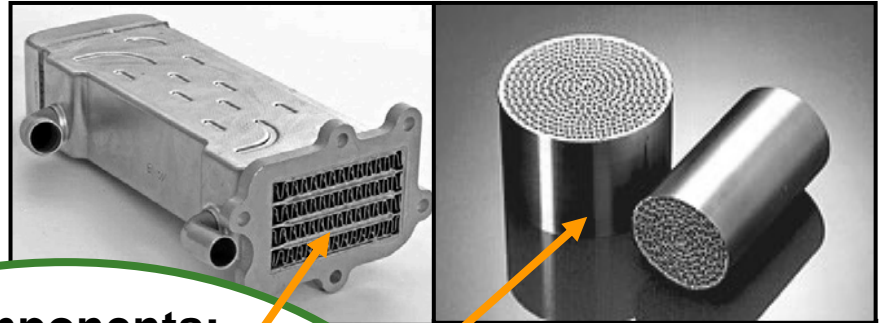
- Compact & Lightweight
- Thin-Walled Sheets
- Metallic Foams
- Feltmetals

## Materials:

- Stainless Steels
- FeCrAl
- Inconels
- Nickel Brazing Alloys

## Components:

**EGR Coolers**  
**Metallic Catalytic Converters**  
**Heat Recovery Systems**  
**Diesel Particulate Filters**  
**etc.**



# Background

## LME Ni Historic Chart (Apr 2004 to Jan 2010)



Ni price jumped up almost 4 times higher since Oct 2005. The peak was June 2007 and price dropped down by the end of 2008.

# Concept of filler metal design

## Fe-Cr-Ni-Mo-Si-P System

- Fe, Cr** Base elements and Cr content must be 20wt% at least to improve corrosion resistance
- Ni** Main element and contains as less as possible with the balance of Fe and Cr
- Mo** Elements added to form fine grain structure to obtain higher strength and improved corrosion resistance
- P, Si** Main elements of temperature depressants

# Developed Fe-Cr based filler metals

Type	Chemical Compositions (wt%)						Solidus °C	Liquidus °C	Brazing Temp. °C
	Fe	Cr	Ni	Mo	Si	P			
IronBraz	27	20	43	-	-	10	1,010	1,065	1,100 – 1,120
TB-2720	27	20	43	-	-	10	1,010	1,065	1,100 – 1,120
TB-3025	30	25	30	2	6	7	1,010	1,065	1,100 – 1,120
TB-3520	35	20	30	2	5	8	1,010	1,060	1,100 – 1,120
TB-4025	40	25	20	2	6	7	1,030	1,085	1,120 – 1,140
TB-4520	45	20	20	2	5	8	1,020	1,080	1,100 – 1,130
TB-5020	50	20	15	2	5	8	1,035	1,100	1,130 – 1,150
TB-5520	55	20	10	2	5	8	1,060	1,115	1,150 – 1,180
TB-6020	60	20	5	2	5	8	1,090	1,130	1,150 – 1,200

# Current Fe-Cr based filler metals

## IronBrazes

Type	Chemical Compositions (wt%)						Solidus °C	Liquidus °C
	Fe	Cr	Ni	Mo	Si	P		
IronBrazes	27 Bal.	20 20.18	43 43.90	-	-	10 8.90	990	1,040
TB-4520	45 Bal.	22 21.91	22 21.76	0.5 0.58	4.5 4.31	6 6.24	1,038	1,051



Foil form of TB-2720

# Other Fe-Cr based filler metals

Product Name	Chemical Compositions (wt%)								Solidus	Liquidus
	Fe	Cr	Ni	Mo	Si	P	Cu	B	°C	°C
VZ2106	35	11	44	1.5	6.4	-	1	1.5	1,044	1,154
VZ2099	51	11.5	29	1.5	2.8	1.8	-	1.9	934	1,146
BrazeLet F300	34	24	20	-	5	7	10	-	B.T.:1,120	
BrazeLet F302	54	15	10	-	7	9	5	-	B.T.:1,100	
Amdry 805	38	29	18	-	7	6	RE:0.2		M.P.:1,104	
AlfaFusion	Bal	17	12	2.2	Mn:1.6		Si+B:?		B.T.:1,190	
FP-641	Bal	18	15	2	5	6.5	2	-	1,030	1,060
FP-642	Bal	18	20	2	2	8	2	-	1,030	1,060
FP-633	20	29	Bal	-	4	6	-	-	1,020	1,060

※RE: Rare Earth

※B.T.: Brazing Temperature

※M.P.: Melting Point



# Current issues

- Price

Not very competitive to conventional Ni based filler metals due to recent Ni price.

- Corrosion resistance

Corrosion resistance characteristic changes depends on corrosion solutions and type of base metals.

- Joint strength

It was found that the value of joint strength fluctuated due to brazing conditions.

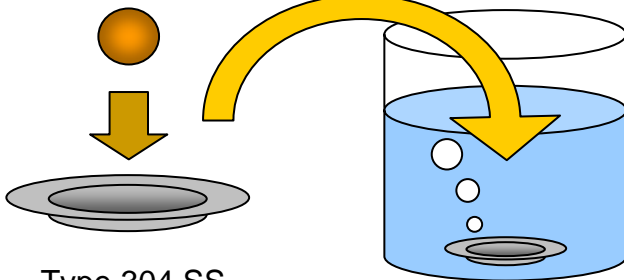
# Corrosion test results (Dipping test)

Appearance and weight loss after 72 hours exposure

Test Solutions	H <sub>2</sub> SO <sub>4</sub> (5%)	NHO <sub>3</sub> (5%)	HCl (5%)	NaClO (5%)	NH <sub>4</sub> OH (5%)
Appearance	No change	No change	Rust	Minute rust	No change
Weight loss (mg)	No change	No change	~10	Unable measure	No change
Results	Excellent	Excellent	Attacked	Good	Excellent

## Dipping test method

TB-4520 500mg



Type 304 SS



Appearance after the test

H<sub>2</sub>SO<sub>4</sub>: Sulfuric Acid

NHO<sub>3</sub>: Nitric Acid

HCl: Hydrochloric Acid

NaClO: Sodium Hypochlorite

NH<sub>4</sub>OH: Aqueous Ammonia

# Corrosion test results (Salt spray test)

Appearance after 168 hours salt spray test

TB-2720



TB-4520



BNi-2

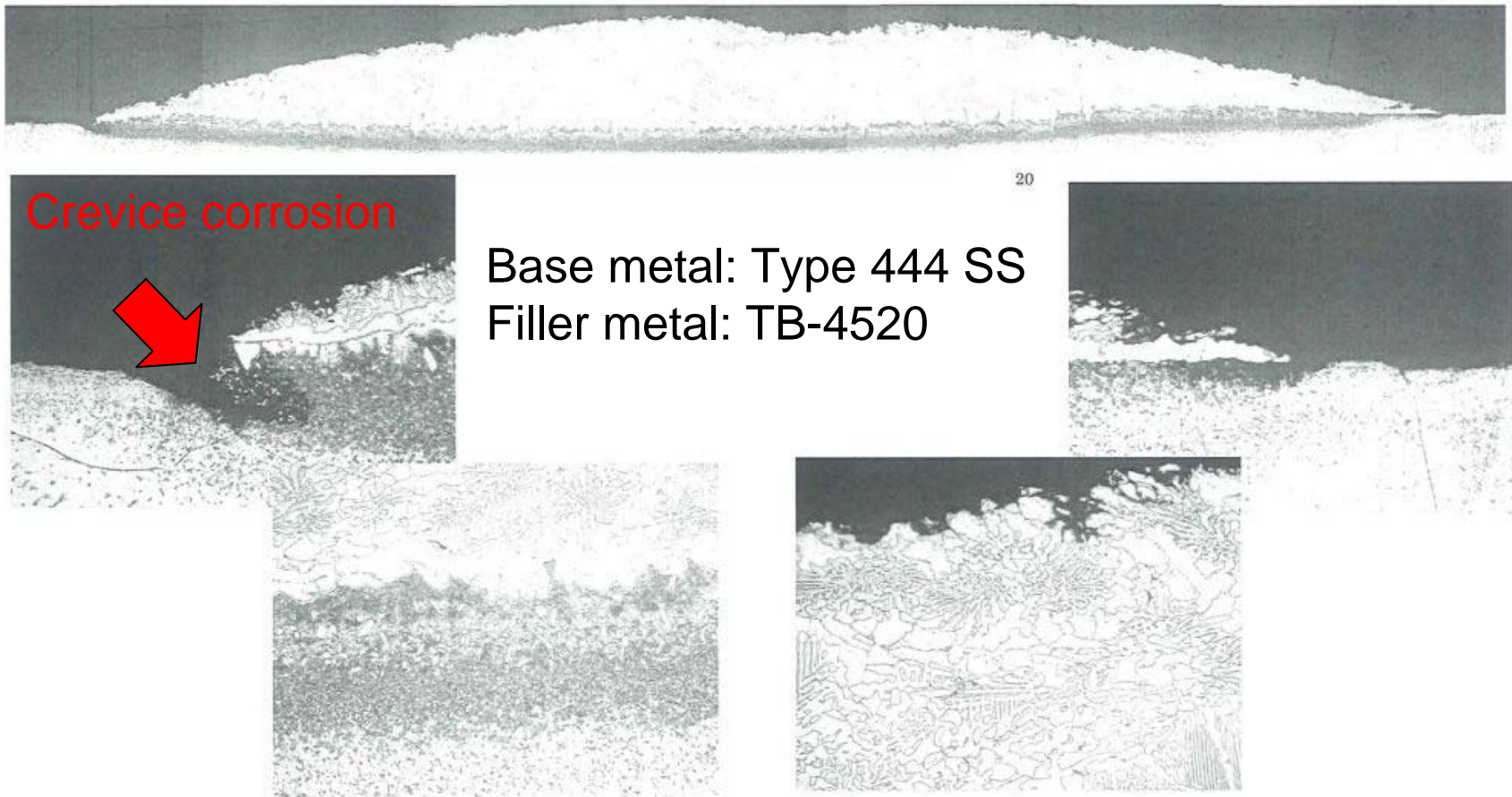


BNi-5



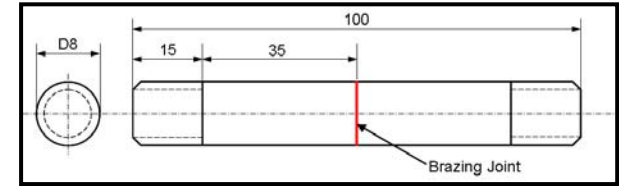
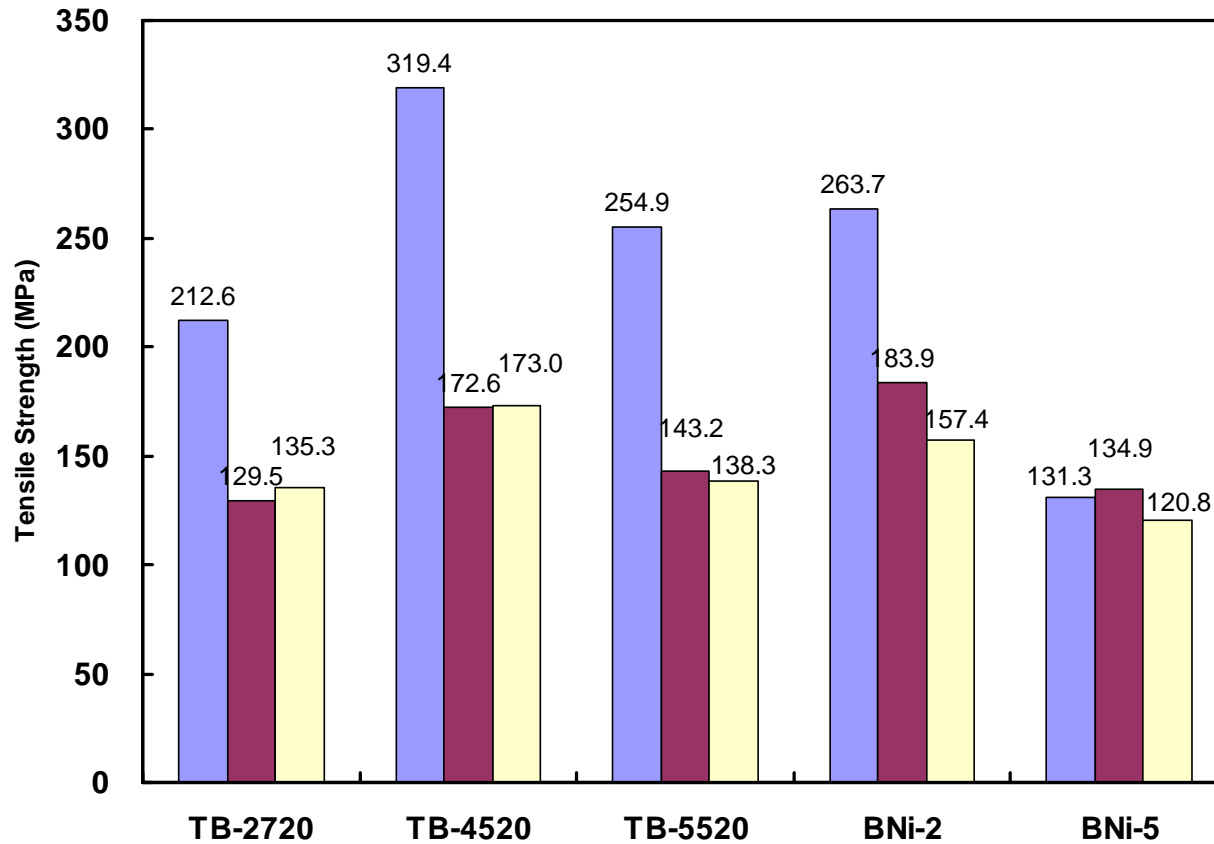
TB-2720 and TB-4520 with type 304 SS show excellent corrosion resistance against to salt spray test.

# Corrosion test results (VDA test)

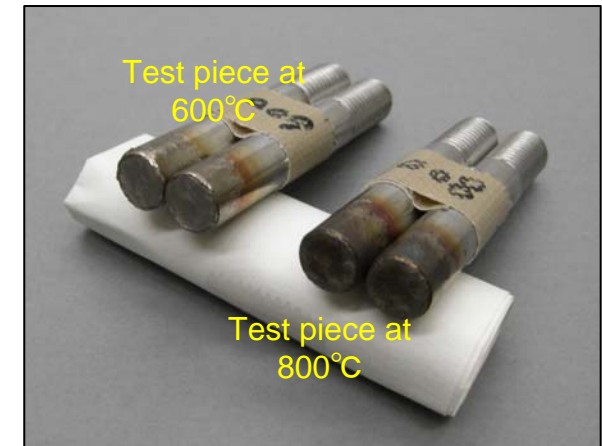
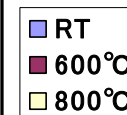


Crevice corrosion at the edge of fillet and base metal on type 444 SS test piece was observed by artificial exhaust gas condensate.

# Joint strength (Tensile test)



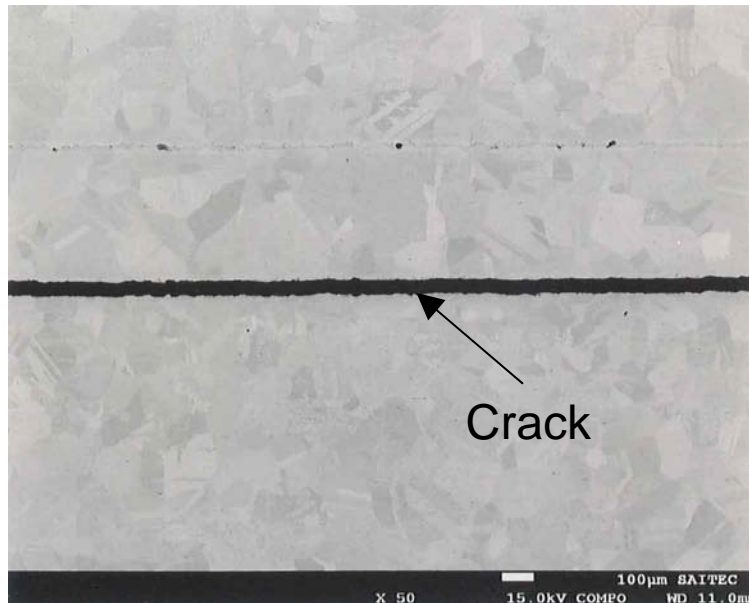
Schematic diagram of tensile test specimen



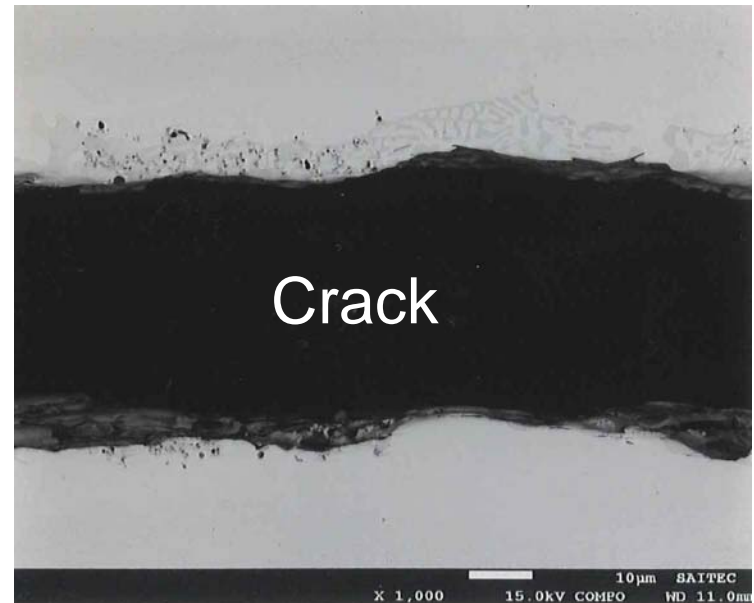
TB-2720 and TB-4520 obtain good joint strength compared to Ni based filler metals.

# Poor joint strength

## Cross-section at the fractured joint



100µm



10µm

Brazing conditions

Load:  $\approx 10\text{kgf/cm}^2$

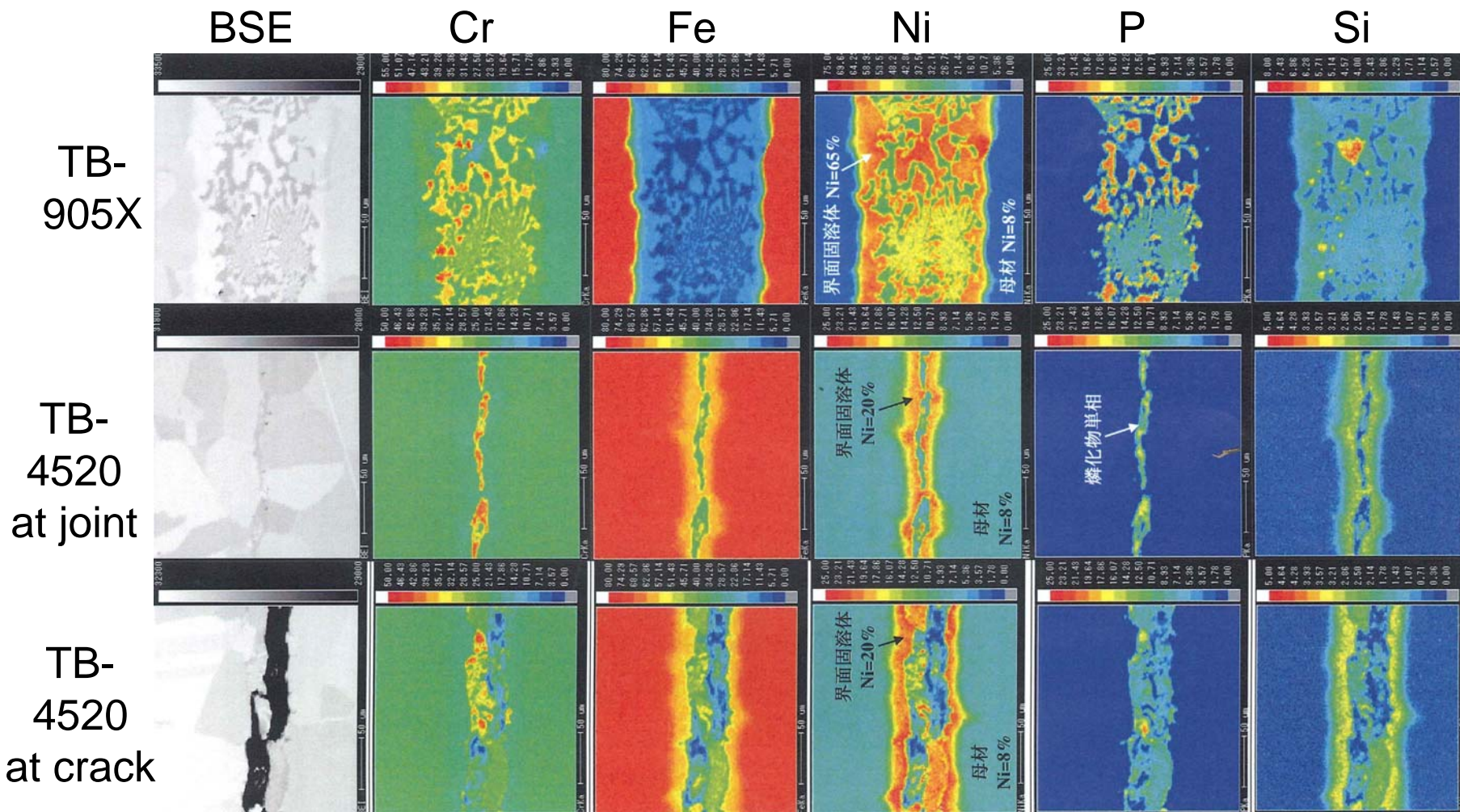
Brazing temp & time: 1,120°C & 60min

Base Metal: Type 316 SS

Brazing filler metal: TB-4520



# Poor joint strength



# Summery

- Fe-Cr based filler metals have good corrosion resistance, but corrosion resistance characteristic changes depends on corrosion solutions and type of base metals.
- Joint strength of Fe-Cr based filler metals fluctuate due to brazing conditions such as long brazing time and heavy load at the joint to form thin brazing layer.
- Fe-Cr based filler metals still need to improve or modify so that to know further basic information is necessary.